

Design Profiles of EATS for Alternate Fuelled Combustion Engines

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- **Introduction & Alternate Fuels In India**
- **Emission Legislation Norms (BS VI & Onwards)**
- **H₂-ICE & Exhaust After treatment System**
- **Blended Ethanol Fuel**
- **Summary & Conclusion**

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Introduction

- Emissions from the Energy sector more than doubled from ~1141 Mt CO₂e in 2005 to ~2455 Mt CO₂e in 2018.
- Transport sector accounted for 12 % of India's total emissions.

Road transport sector – One of the most energy intensive sectors.

- Total energy demand from road transport is projected to double from 2023 to 2050

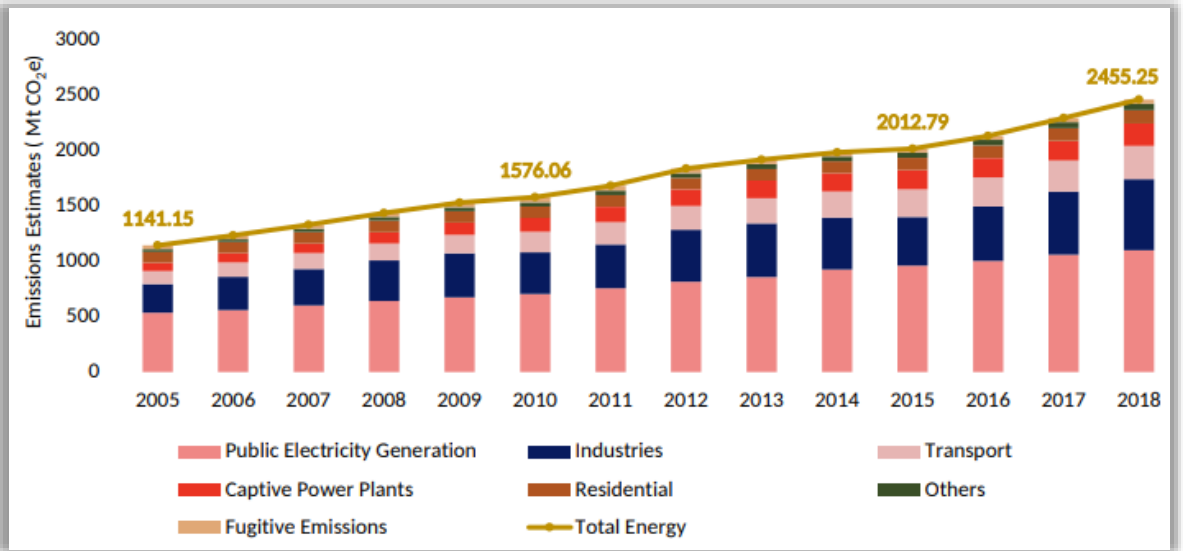
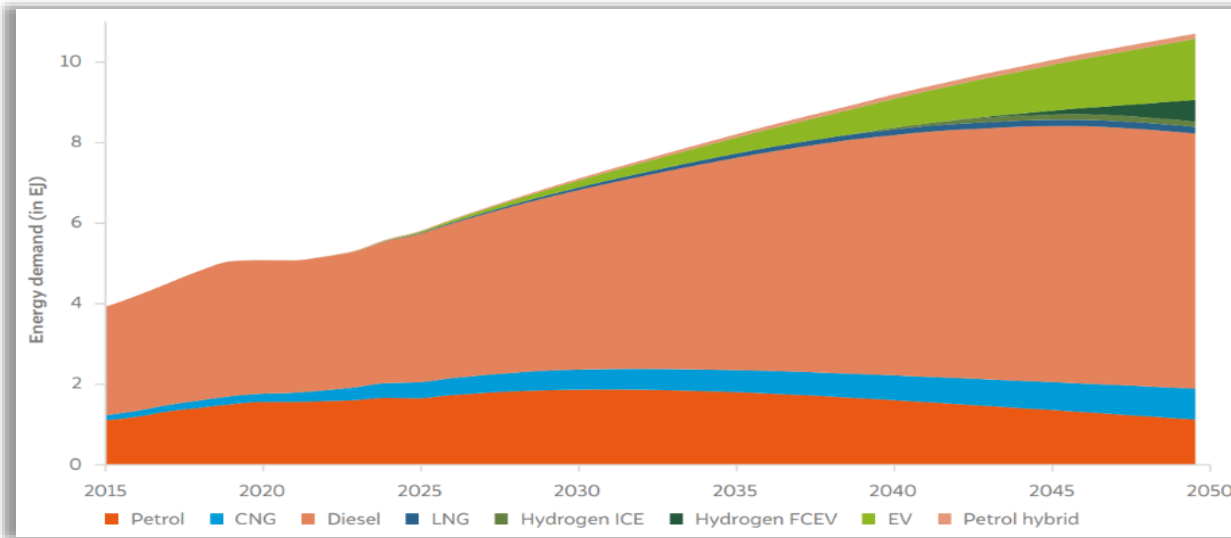
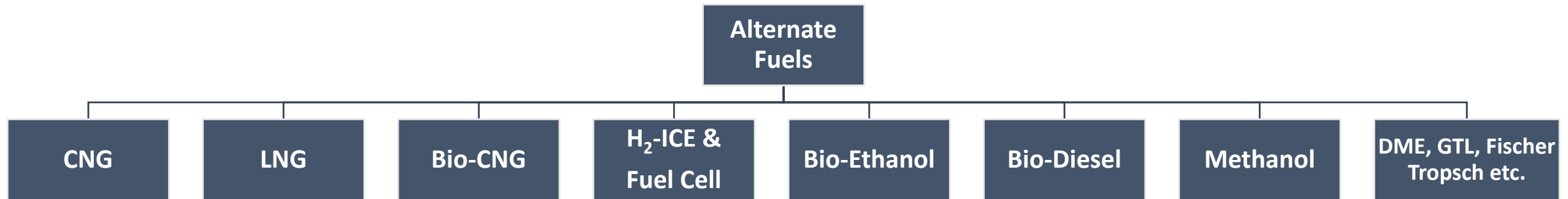


Fig 1: Category-wise Emissions (Mt CO₂e) and Percentage Share in Total Energy Sector Emissions (2018)

“The need for Alternate Fuels arises from the urgent global demand to reduce environmental pollution, enhance energy security, and create a sustainable future beyond fossil fuels.”



Need for Alternate fuels in India

- Import dependency – Energy security risks.
- Environmental impact
- Resource depletion
- Energy Diversification
- Rural & Agricultural Linkages

Key Challenges in Adoption

- High Upfront Costs
- Technological Readiness
- Fuel Supply & Availability
- Vehicle Compatibility
- Consumer Awareness & Acceptance

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🔮 Looking Ahead: BS7

Assuming BS7 adopts Euro 7 framework.

Key Changes

- Tightened Emission limits.
- Extended Lifetime (Up to 200,000 km for LDV & up to 875,000km for HDV)
- On-Board Monitoring system
- Brake & Tyre Emission limits.

Light Duty Vehicles

- Alignment with Euro 6e (Emission limits & Test methods).
- PN10 instead of PN 23.

Euro 7: Outcome of the final Trilogue(Council of the EU)-HDV

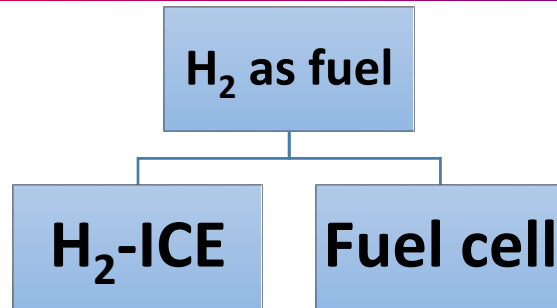
M2, M3, N2, N3	Euro-6		Euro-7
mg/kWh (gas) kWh-1 (PN)	WHTC	WHTC(CI & PI)/ WHSC(CI)	RDE
NO _x	460	200	260
PM	10	8	10
PN	PN ²³ - 6x10 ¹¹	PN ¹⁰ - 6x10 ¹¹	PN ¹⁰ - 9x10 ¹¹
CO	4000	1500	1950
NMOG	160 _{THC}	80	105
NH3	10ppm	60	85
CH4	500	500	650
N ₂ O	-	200	260

M2: busses<5 t,M3: busses>5 t, N2: trucks 3.5 t – 12 t , N3: trucks > 12 t

Heavy Duty Vehicles

- NO_x emissions reduced by 56%
- CH₄, N₂O, PN₁₀, NMOG added to the regulated emissions
- New RDE requirements

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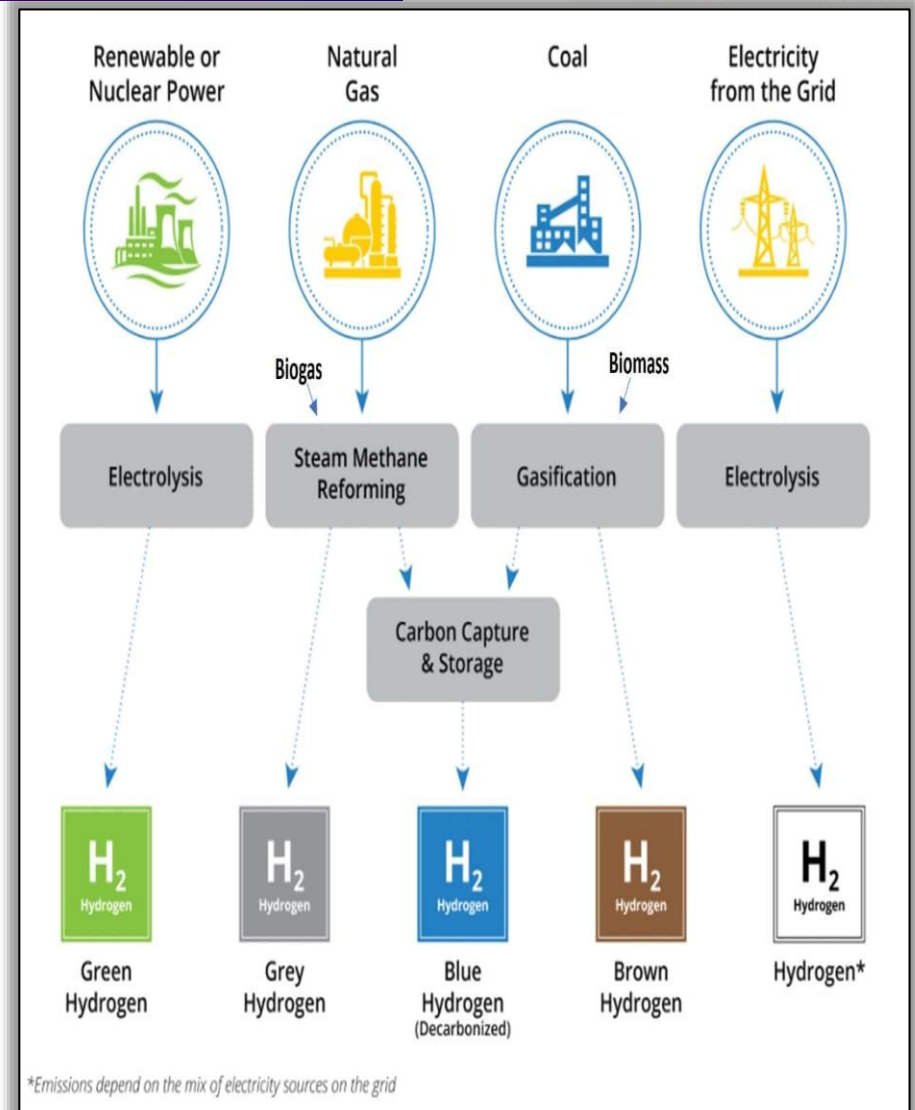


- Based on hydrogen combustion
- Adaptation to existing ICE technology
- Lower H₂ purity needed
- NOx emissions, H₂ slip, and water vapor

- Based on electrochemical reaction
- New Infrastructure needed
- Hydrogen purity required
- No other emission than water

H₂-ICE technology is favoured for easier integration and lower cost than fuel cells.

- Sustainable Pathway to Zero Emissions and Decarbonization.
- ICE technology allows integration of this fuel.
- Better performance with lower emissions.



**Emissions depend on the mix of electricity sources on the grid*

Fig 2: The different methods of hydrogen production and color-coding system (Hydrogen Power Partners)

H₂ as fuel

PROPERTIES	UNITS	HYDROGEN	GASOLINE	METHANE
Density gaseous	[kg m ⁻³]	0.08	4.4	0.65
Diffusivity in air	[cm ² s ⁻²]	0.63	0.08	0.20
Flammability in air	[volume %]	4.0 – 75.0	1.0 – 7.6	5.3 – 15.0
Minimum energy for ignition in air	[mJ]	0.02	0.24	0.29
Auto-ignition temperature	[K]	858	501-744	813
Flame temperature in air	[K]	2318	2470	2148
Burning velocity in NTP air	[cm s ⁻¹]	265 – 325	37 – 43	37 – 45
Quenching gap in NTP air	[mm]	0.64	2.00	2.03

Favorable properties enabling high combustion efficiency

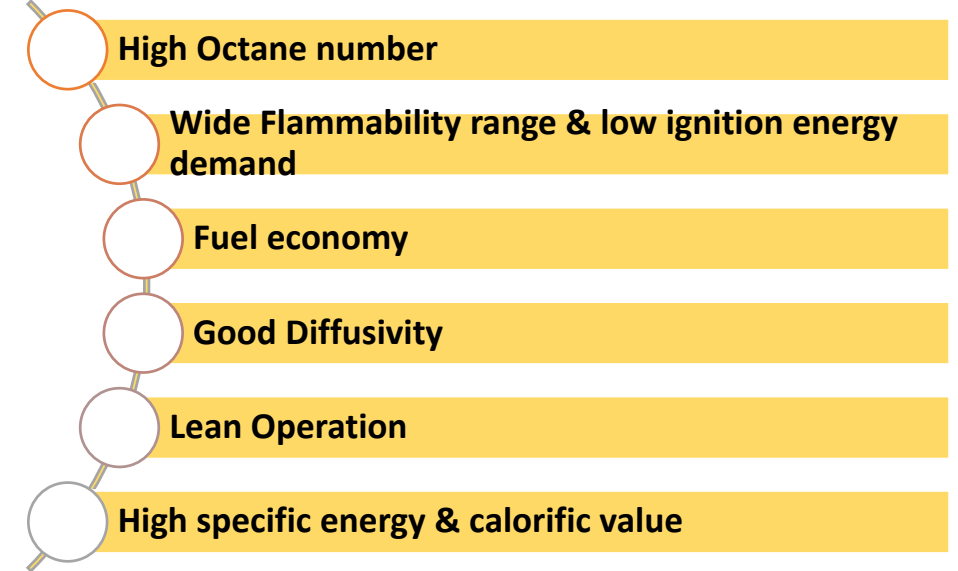


Fig 3: Hydrogen Properties

- Standalone fuel in both SI and CI engines.
- High stoichiometric air-to-fuel ratio (34.4:1)
- Lean operation due to wide flammability range allowing for NO_x control.

- Low energy of ignition-Premature, Uncontrolled ignition
- High auto ignition temperature (SI over compression ignition)
- High Antiknock properties

H₂-ICE emission characteristics & challenges

- Significant amount of Hydrogen, Water & NO_x.
- Varying NO_x emissions based on engine calibration.
- Exhaust Temperature slightly lower than Diesel ICEs.
- High amount of water production (~ 30%).
- CO, HC & PM from lube oil combustion.

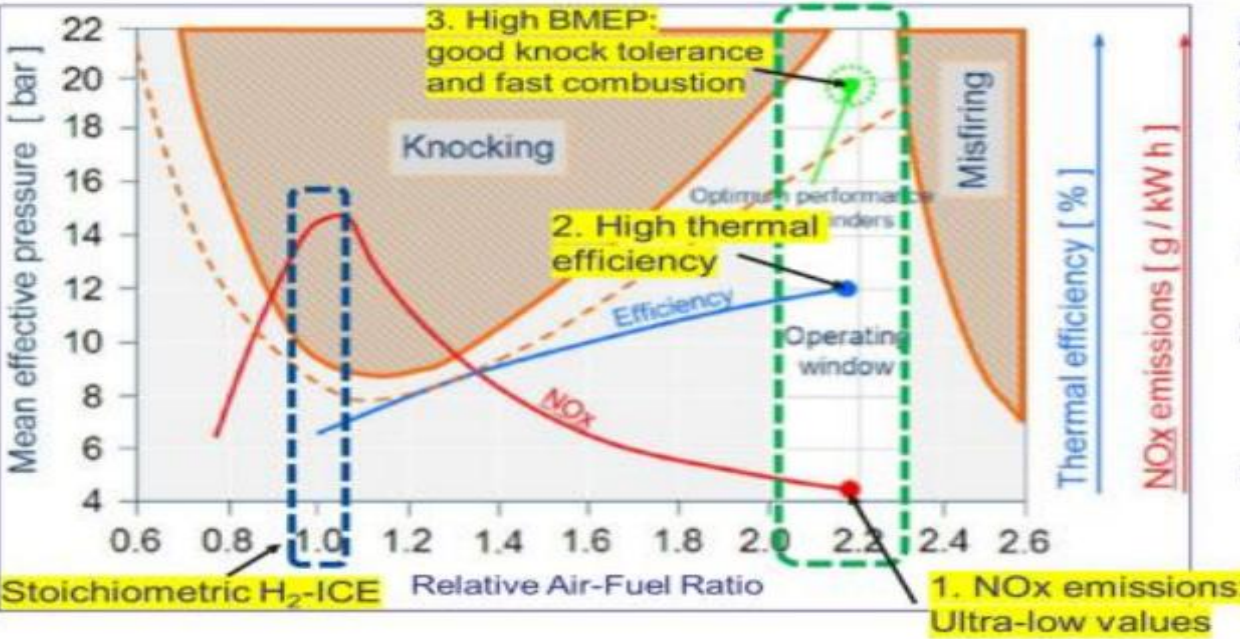


Fig 4: Relationship between AFR, mean effective pressure, thermal efficiency & NO_x emissions

TECHNOLOGY	DIESEL ENGINE	PFI SI	LP DI SI	HP DI CI
Fuel	Diesel	H ₂	H ₂	H ₂ /Diesel
Injection/ignition type	DI/CI	PFI/SI	DI/SI	DI/CI
Air/fuel ratio	1.1 < λ < 1.65	1.8 < λ < 2.7	1.8 < λ < 2.7	1.5 < λ < 2.4
Engine-out NO _x	High	Low	Medium	High
After-treatment requirement(Euro VI)	Yes	No	Probably	Yes
H ₂ O in exhaust	Reference	3.1 times more	N.A.	2.6 times more
Exhaust temperature	Reference	Lower	Lower	Equal
Expected market Introduction	-	2027	2028	2025-2027
Power density	Reference	Lower	Equal	Higher
Efficiency	~45	~42	~42	>45

Fig 5: H₂-ICE technologies compared with the traditional diesel engine

Challenges

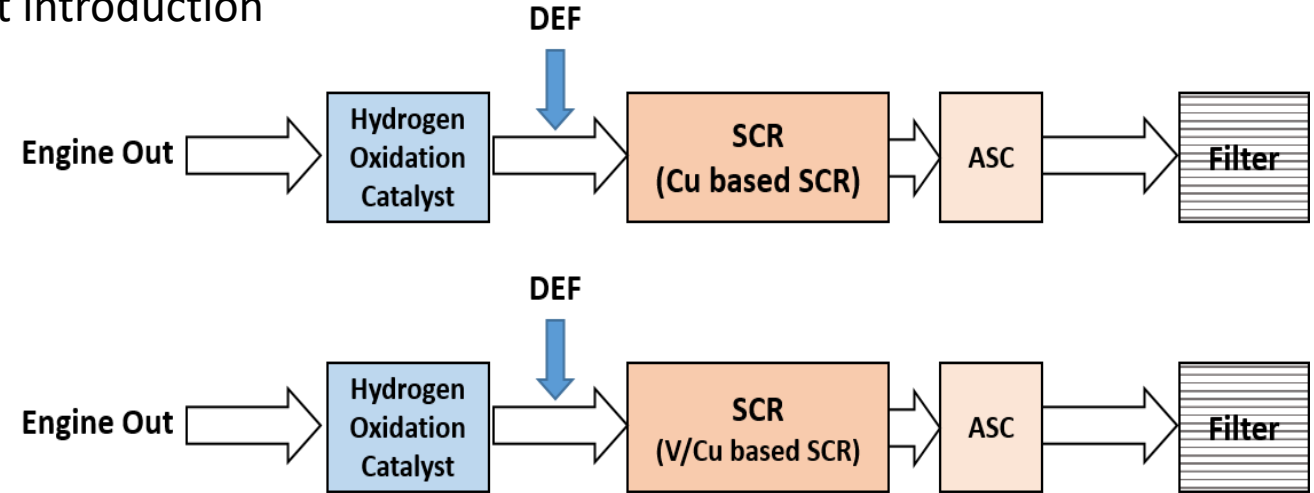
- High thermal efficiency with reduced emissions.
- Stability of the combustion process.
- High specific engine power.
- Right Injection strategy (PFI, MPI or DI).
- Engine knock, pre-ignition & ignited mixture backdraft.

H₂-ICE Exhaust Aftertreatment system

OEM's perspective: EATS development for quick market introduction

Ecocat H₂-ICE ATS configuration for lean operation

High DeNO_x efficiency
Low N₂O formation
Wider Temperature Window



Reducing harmful emissions

- Existing SCR technology for efficient NO_x removal (0.25-6g/kWh)
- Oxidation Catalyst for H₂-slip.
- ASC for ammonia slip

Temperature window – 200-500°C

Water content-0-30%

Reactions in H₂-ICE Exhaust After treatment System

	$2 \text{H}_2 + \text{O}_2 \rightleftharpoons 2 \text{H}_2\text{O}$	
	$\text{NO} + \text{O}_2 \rightleftharpoons \text{NO}_2$	HOC catalyst
Hydrolysis	$\text{CO}(\text{NH}_2)_2 + \text{H}_2\text{O} \rightleftharpoons 2\text{NH}_3 + \text{CO}_2$	
Standard SCR	$4 \text{NO} + 4 \text{NH}_3 + \text{O}_2 \rightleftharpoons 4 \text{N}_2 + 6 \text{H}_2\text{O}$	
Fast SCR	$\text{NO} + 2 \text{NH}_3 + \text{NO}_2 \rightleftharpoons 4 \text{N}_2 + 3 \text{H}_2\text{O}$	SCR catalyst
NO ₂ -SCR	$4 \text{NH}_3 + 3 \text{NO}_2 \rightleftharpoons 3.5 \text{N}_2 + 6 \text{H}_2\text{O}$	
	$4 \text{NH}_3 + 3 \text{O}_2 \rightleftharpoons 2 \text{N}_2 + 6 \text{H}_2\text{O}$	ASC catalyst

H₂-ICE Exhaust Aftertreatment system

Key performance matrix

- **Oxidation catalyst:** High H₂ Oxidation performance, NO₂ formation & Exotherm.
- **SCR:** High NO_x removal percentage
Low N₂O formation
- **Hydrothermal durability of catalysts (0-30%).**

Right Sizing of Catalyst based on application meeting limits.

Influence Parameters

1. NH₃ storage capacity of SCR
2. Effect of H₂ on SCR Performance
3. Effect of NO₂ to NO_x ratio
4. Effect of water on oxidation catalyst & SCR performance
5. Effect of Space Velocity on catalyst performance

SCR Washcoat Technology– NO_x abatement

Copper based SCR

- Wide temperature operation
- High Thermal Durability
- High Temperature Conversion

Vanadium based SCR

- Low temperature range
- Low Thermal Durability
- Low Temperature Conversion

Challenges involving SCR technology for H₂-ICE

Low & high temperature NO_x conversion

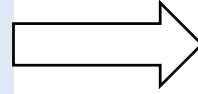
Effect of moisture content on SCR durability

Increased system complexity

DeNOx & Catalyst Durability

NH₃- Temperature Programmed Desorption Study

- Acidic properties crucial for NH₃ adsorption
- NH₃ adsorption capacity



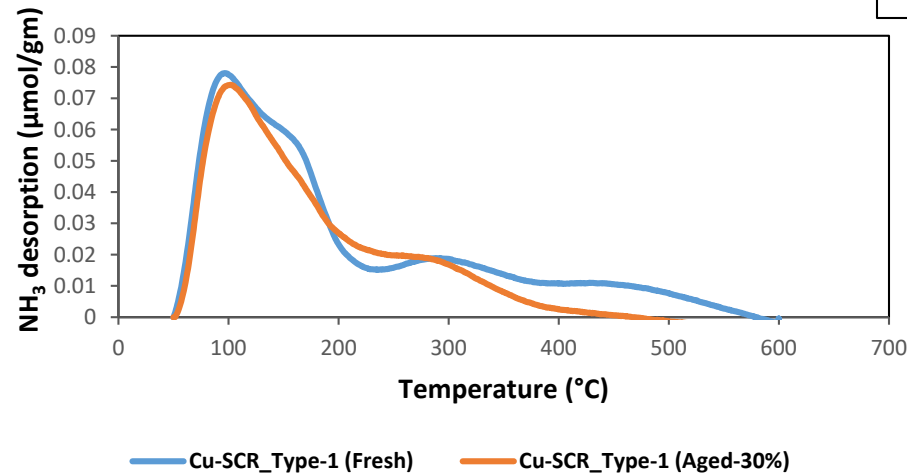
In SCR reaction, active site has a strong adsorption effect on the gas molecules

➤ **Directly relates to NOx conversion**

➤ **Hydrothermal Aging - 650°C x 50hrs, 30% H₂O**

Type-1: High Silica
Type-2: Low Silica

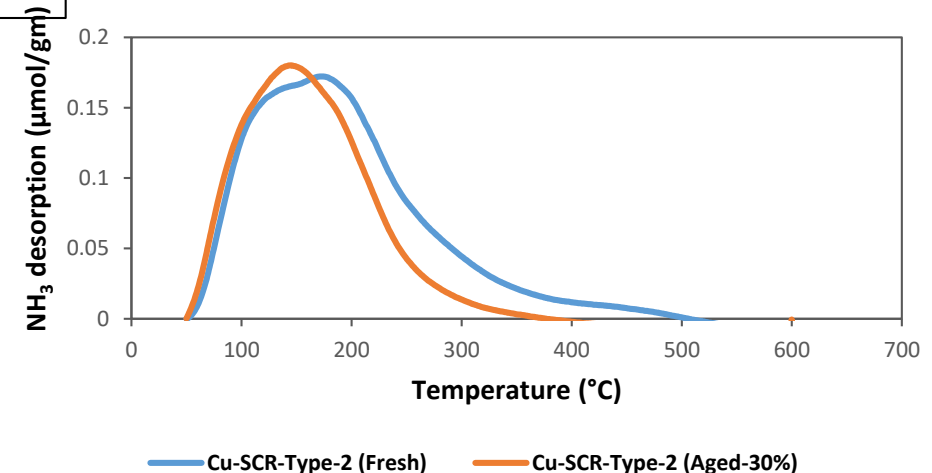
Cu-SCR_Type-1 (Fresh & Aged)



Cu-SCR_Type-1

- High Temperature operating range.
- Reduced Low Temperature NH₃ storage capacity.
- Stable under severe Hydrothermal Aging.
- Good balance between activity & stability.

Cu-SCR_Type-2 (Fresh & Aged)



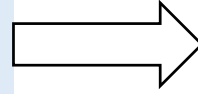
Cu-SCR_Type-2

- Low to mid temperature operating range.
- High low Temperature NH₃ storage capacity.
- Tends to suffer from Hydrothermal aging

DeNOx & Catalyst Durability

NH₃- Temperature Programmed Desorption Study

- Acidic properties crucial for NH₃ adsorption
- NH₃ adsorption capacity

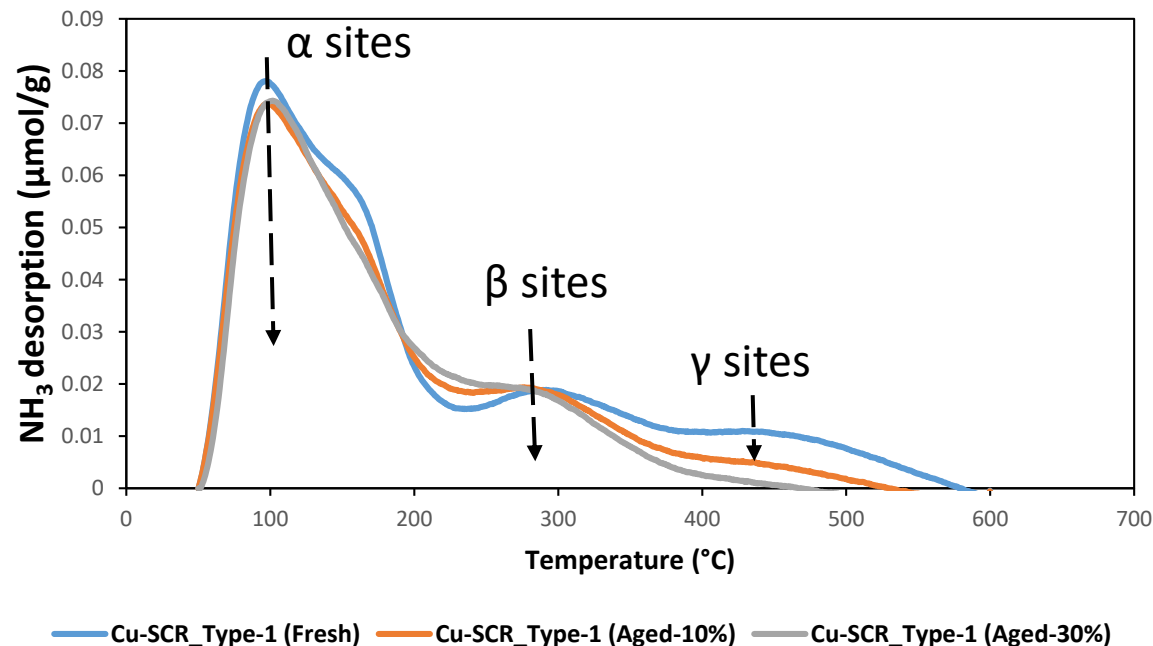


NH₃ adsorbs on

1. α sites-Weak Lewis acid sites
2. β sites-Strong Lewis acid sites
3. γ sites-Bronsted acid sites

- HT Aging- 650°C x 50hrs, 10% H₂O – Diesel Aging
- HT Aging- 650°C x 50hrs, 30% H₂O – H₂-ICE Aging

Cu-SCR Type-1 (Fresh vs Aged)



Effect of Hydrothermal Aging (Fresh, 10%, 30%)

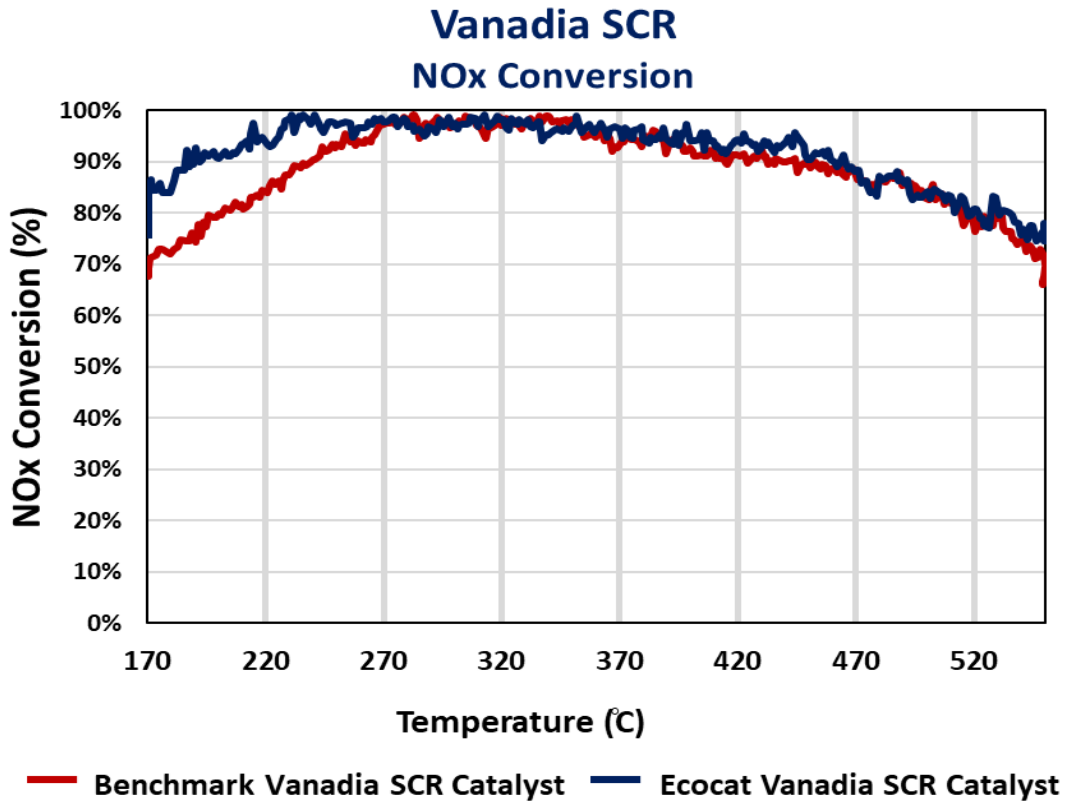
- NH₃ adsorption capacity is retained.
 - Unstable ZCuOH active sites $\leftrightarrow$ more stable Z₂Cu sites.
 - Lewis Sites maintained even after severe aging.
 - Significant loss of γ -sites (Bronsted acid sites).
- Dealumination of the framework**
- Catalyst performance remains the same.

SCR Washcoat Technology – High & Low Temperature NOx conversion



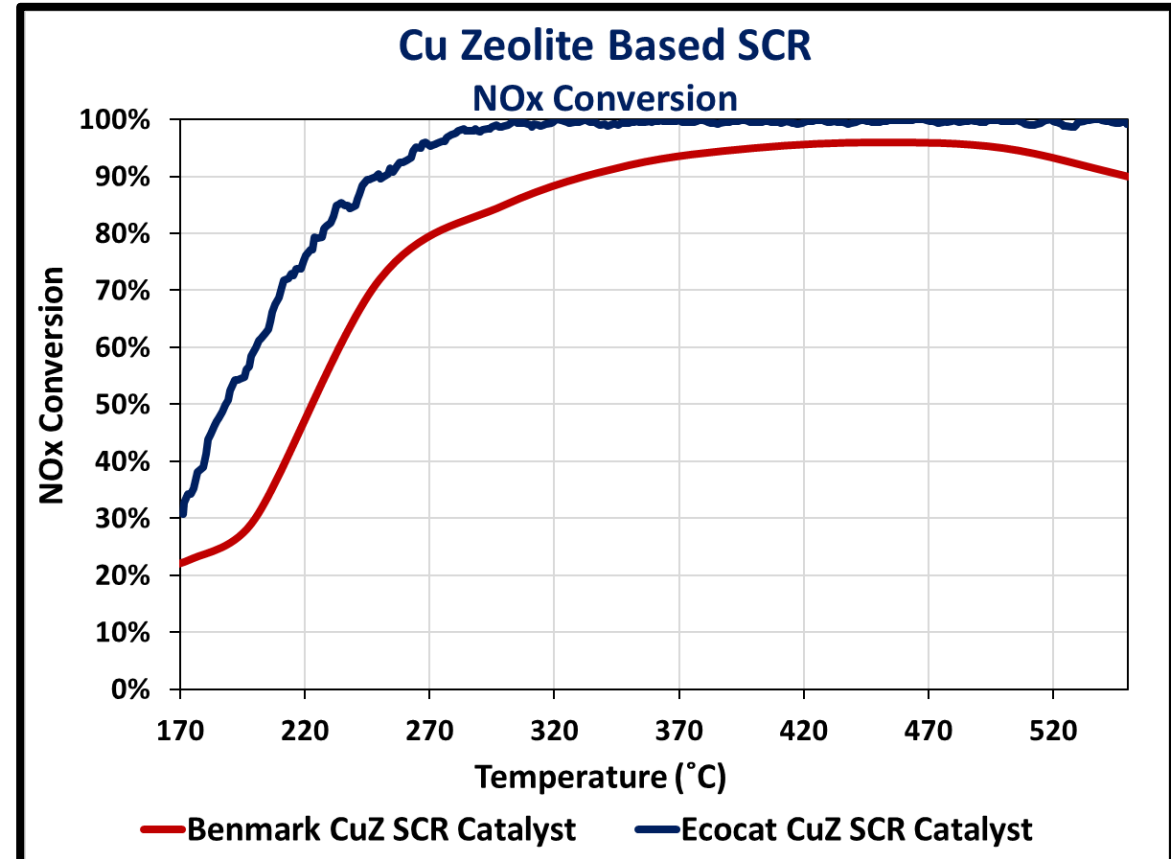
Low & Medium Temperature Technology

Vanadia SCR



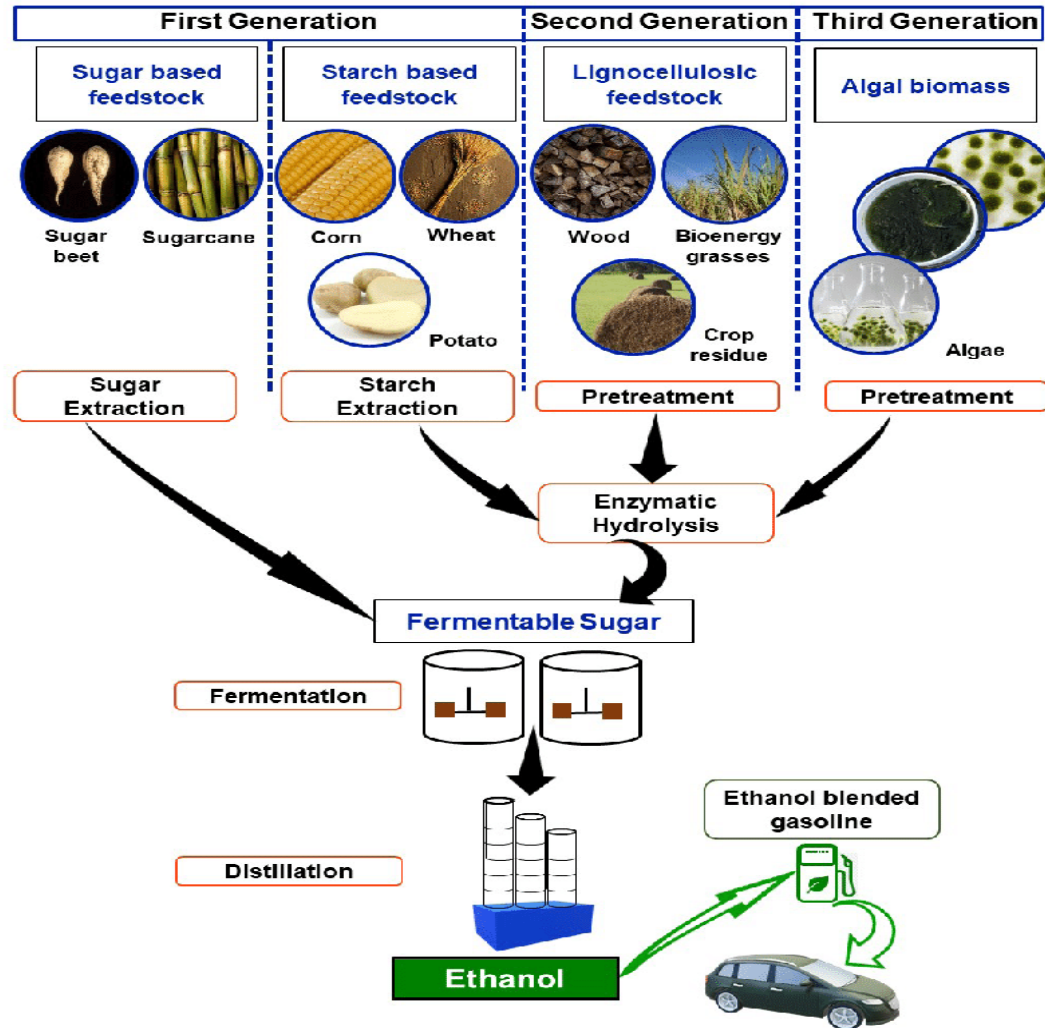
Wider Temperature Technology

Cu-Zeolite SCR



- SCR washcoat chemistry shows efficient NOx conversion at wider temperature range with lower LOT.

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India's Focus

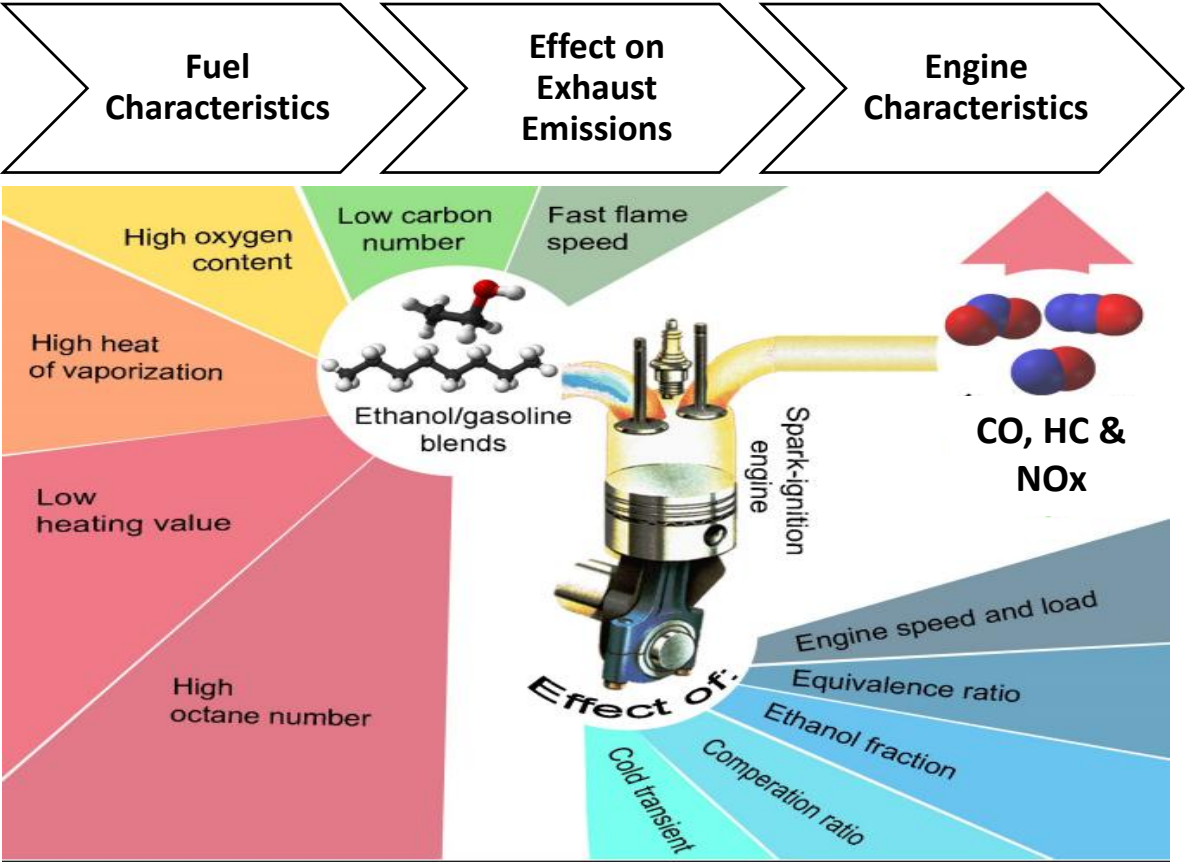
India mainly uses **1G ethanol**

- The National Policy on Biofuels is now promoting **2G ethanol**
- **BPCL's first integrated 2G plus 1G Bio Ethanol refinery at Bargarh**
- **IOCL: 2G Ethanol Plant set up using rice straw & 3G Ethanol Technology with Lanzatech gas fermentation technology at Panipat Refinery**

- Ethanol blended gasoline in SI engines is accepted worldwide.
- Measurable GHG emissions benefits as vehicular fuel.
- Favorable Exhaust Emission reducing properties.
- Flex fuel vehicles with E-85 in Brazil & USA currently in use.

Fig: 6 Schematic pathways of first, second and third generation ethanol production

Ethanol



- Ethanol blending increases the oxygen content.
- Increase in octane number & the volatility of gasoline.
- Lower Energy density – Fuel consumption.
- Minor tailpipe CO₂ reduction.

Property	Gasoline	Ethanol	Remarks
Stoichiometric AFR[-]	14.2-15	9	More Power Potential
O-fraction [mass%]	0	34.7	More complete combustion
Research octane number	91-100	110	High compression ratio, High thermal Efficiency, More Power
Reid vapor pressure [kPa]	53-60	17	Difficulty In cold start, High HC & CO during cold start
Lower Heating value [MJ/kg]	44	27	Higher Fuel Consumption
Latent Heat vaporization [kJ/kg]	380-400	910	Faster combustion and higher volumetric efficiency

Fig: 7 Physiochemical properties of Gasoline & Ethanol

Effect of Ethanol-gasoline blends

Based on literature

Regulated gasoline(RF) compared with E15 & E30 blended fuels

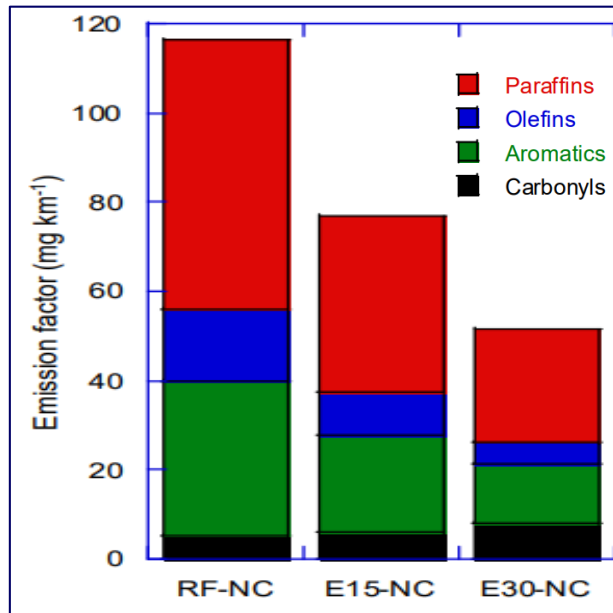


Fig: 8 VOCs emission factors for RF, E15, E30

RF- 116mgkm⁻¹, E15- 76.8mgkm⁻¹, E30- 51.5mgkm⁻¹

➤ Regulated gases emissions

CO, THC, NMHC, NO_x, CO₂

➤ Unregulated gases Emissions

Air toxics emissions: HCHO, CH₃CHO, NH₃, CH₃CH₂OH

GHG emissions: N₂O, CH₄

An analysis of the VOC species in the tailpipe exhaust

• Addition of ethanol

Paraffins ↓, Olefins ↓, Aromatics ↓, Carbonyls ↑

- **Carbonyls emissions:** acetaldehyde, acetone, formaldehyde, and benzaldehyde
- **HC emissions tend towards low molecular weight compounds.**
- **Low exhaust temperature-cooling effect of Ethanol**

Effect of Ethanol-gasoline blends

Based on literature

Regulated gasoline(RF) compared with E15 & E30

CO emission- ↓ 30%(E15), ↓ 37%(E30)

HC emission- ↓ 19%(E15), ↓ 28%(E30)

NO_x emission- ↑ 8.1%(E15), ↓ 2.7%(E30)

- Ethanol blended gasoline reduces CO & HC emissions from tailpipe exhausts
- Aldehydes (formaldehydes & acetaldehydes), unburned ethanol emissions increases
- NO_x emission depending on operating conditions and other parameters

Aging effect on TWCs with gasoline-ethanol blend

Conventional TWC: PGM on Al₂O₃, CeO₂-ZrO₂ & other dopants (5000 → 16200km)

CO: (RF-6.1→25%), (E15-5.3→12%), (E30-81→104%)

HC: (RF-9.4→25%), (E15-13→52%), (E30-4.8→29%)

NO_x: (RF-14→43%), (E15-33→83%), (E30-14→57%)

Increase in fuel consumption with Ethanol blending due to low energy content of Ethanol

Effects on the performance of motorcycle catalysts & emissions

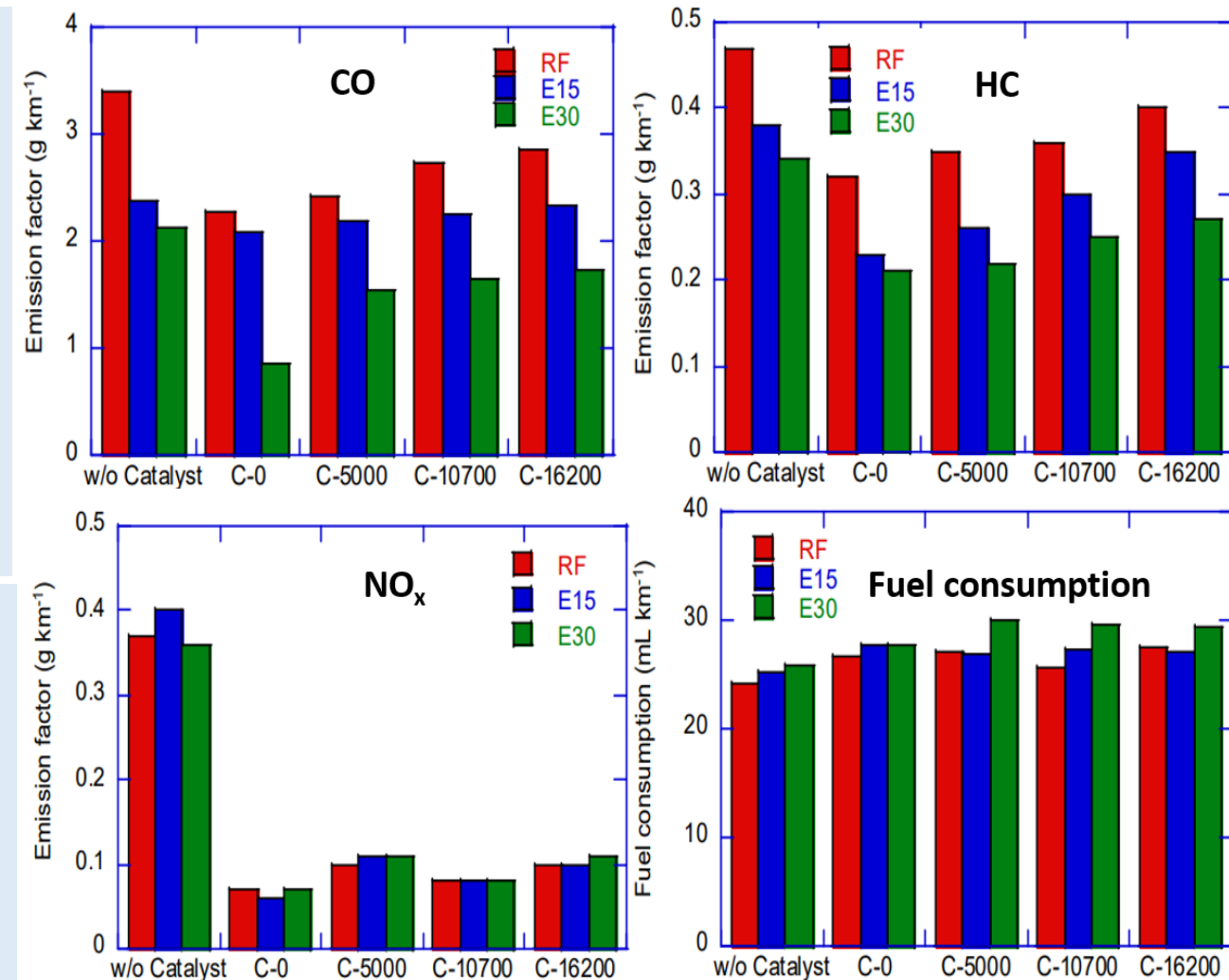


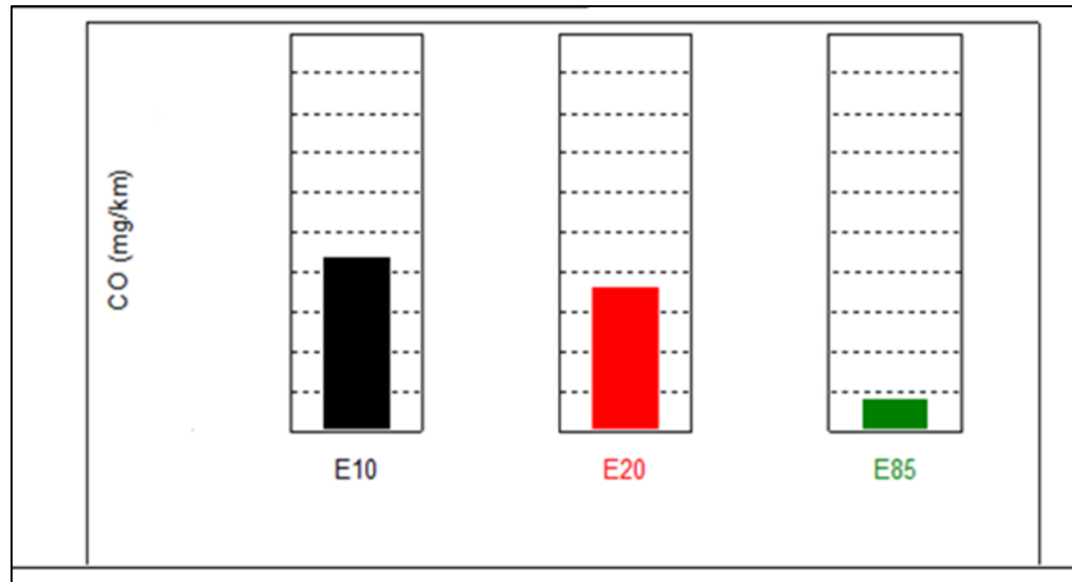
Fig 8: CO, HC, NO_x emission & fuel consumption for different ethanol blending with & without catalyst.(running mileage: 0, 5000, 10700 & 16200km)

Effect of Ethanol-gasoline blends

Effects on the performance of motorcycle catalysts & emissions

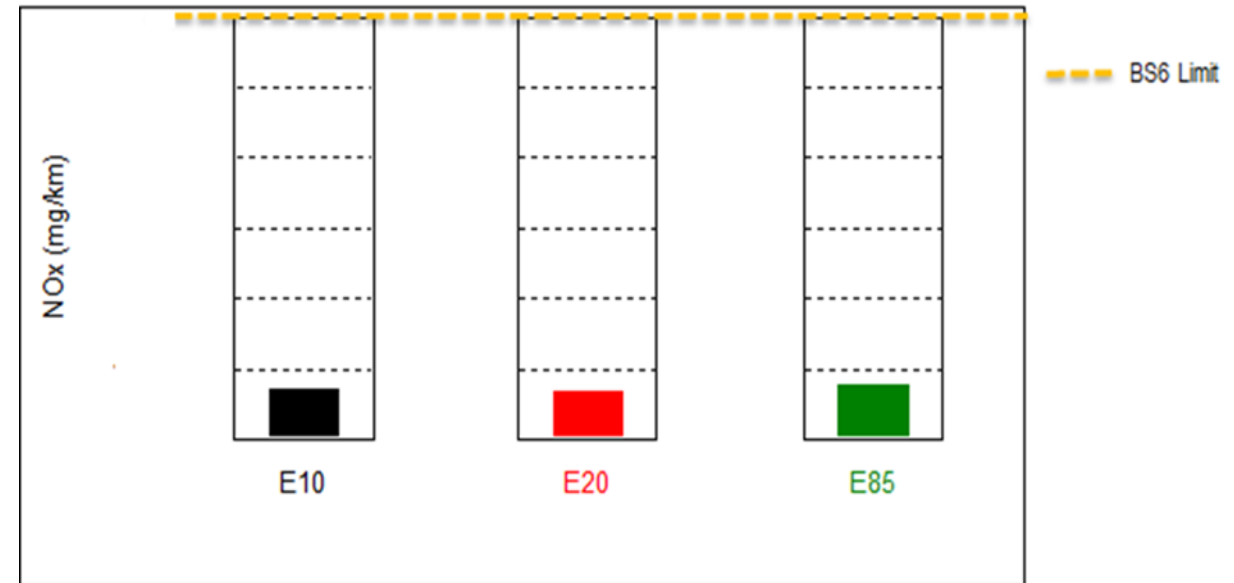
Experimental: All comparisons are between E20 and E85.

Impact on emissions: CO ↓



- Significant reduction in CO observed with increase in ethanol % in blend.
- Overall CO reduced by 77% from E20 to E85.

Impact on emissions: NOx =

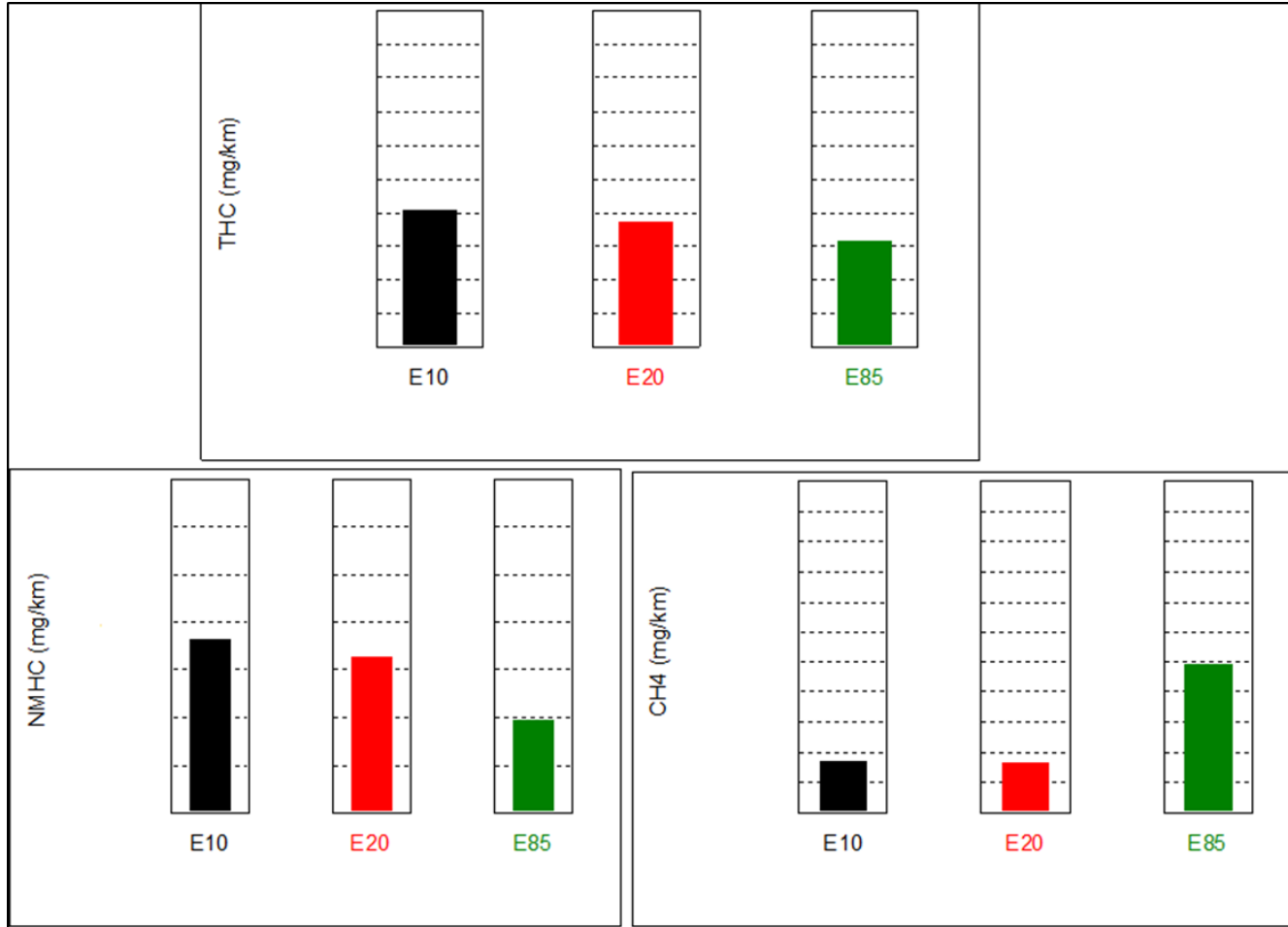


- NOx emission is in similar level with all the tested fuel blends, with the help of the calibration refinement.

Effect of Ethanol-gasoline blends

Effects on the performance of motorcycle catalysts & emissions

Experimental: All comparisons are between E20 and E85.



Impact on emissions: THC / NMHC ↓

- Improvement to the extent of ~15% in THC emissions observed with higher blend.
- Significant improvement, up to ~40% in NMHC emissions observed with increasing ethanol % in blend.

1. NH_3 Desorption studies conducted to understand retention of acid sites, which directly relates to NO_x conversion on SCR.
2. After Hydrothermal aging with 30% steam, Cu based SCR sustained its performance.
3. SCR washcoats cover a wide range of temperatures for NO_x conversion.
4. Emission trends for Ethanol blended Gasoline fuel is discussed.

THANKS