# HC-SCR of NOx for light duty diesel vehicles

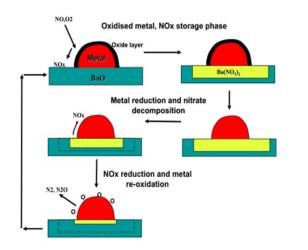
CSIR-NCL

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# Strategies for diesel car aftertreatment

NOx Storage Reduction (NSR) 2NO +  $2O_2^{\text{Pt}} \rightarrow 2NO_2 + O_2^{\text{Ba}} \rightarrow Ba(NO_3)_2$ 

 $\text{Ba(NO}_3)_2 + \text{HC} + \text{H}_2 + \text{CO} \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{N}_2$ 



Developed technology

- Optimised catalyst composition
- $\sqrt[9]$  Need expensive Pt for NO  $\Rightarrow$  NO<sub>2</sub>
- Permanent deactivation of storage component (Ba) even with traces of S
- Needs to replace catalyst after complete formation of BaSO<sub>4</sub>

NH<sub>3</sub> (Urea) - SCR 4NH<sub>3</sub> + 4NO +  $O_2 \rightarrow 4N_2 + 6H_2O$ 

V<sub>2</sub>O<sub>5</sub>-WO<sub>3</sub>/TiO<sub>2</sub>, Zeolite based

- Developed technology
  Suitable for heavy duty vehicles or stationary sources
- Need to carry urea tank on board
- Not suitable for small vehicles due to space constraints

**Opportunity for** 

development

Selective Catalytic Reduction (SCR)

 $\begin{array}{c} \textbf{HC-SCR} \\ \textbf{2CO} + \textbf{2NO} \rightarrow \textbf{2CO}_2 + \textbf{N}_2 \\ \textbf{HC} + \textbf{NO} \rightarrow \textbf{CO}_2 + \textbf{H}_2\textbf{O} + \textbf{N}_2 \\ \textbf{2H}_2 + \textbf{2NO} \rightarrow \textbf{2H}_2\textbf{O} + \textbf{N}_2 \end{array}$ 

- Unburnt hydrocarbon or small fuel penalty can be used for NOx reduction
- Specially suitable for small vehicles
- No need to carry additional tank for reductant

Ready technology not available

Initial work carried out by NCL

### HC-SCR of NOx under lean conditions

Advantages compared to Urea-SCR

- ✤ No need to carry additional urea tank
- ✤ No problem of NH<sub>3</sub> slip
- Use of un-burnt HC for reduction of NOx or slight

additional fuel penalty

Preferred for passenger cars and small carrier vehicles

#### Work Carried Out So Far .....

#### <u>**2 wt% Ag/Al<sub>2</sub>O<sub>3</sub>**</u> is used as benchmark catalyst for HC-SCR

- ↑ Maximum NOx reduction activity
- Maximum selectivity for N<sub>2</sub>
- ✤ Poor activity at lower (<200 °C) and higher temperature (>400 °C)
- Low sulfur tolerance
- Moderate activity in presence of water

Activity can be improved by

- Modification of support or
- Addition of second metal

Catalysts compositions developed and evaluated at CSIR-NCL

Ag-Au/Al<sub>2</sub>O<sub>3</sub> catalysts

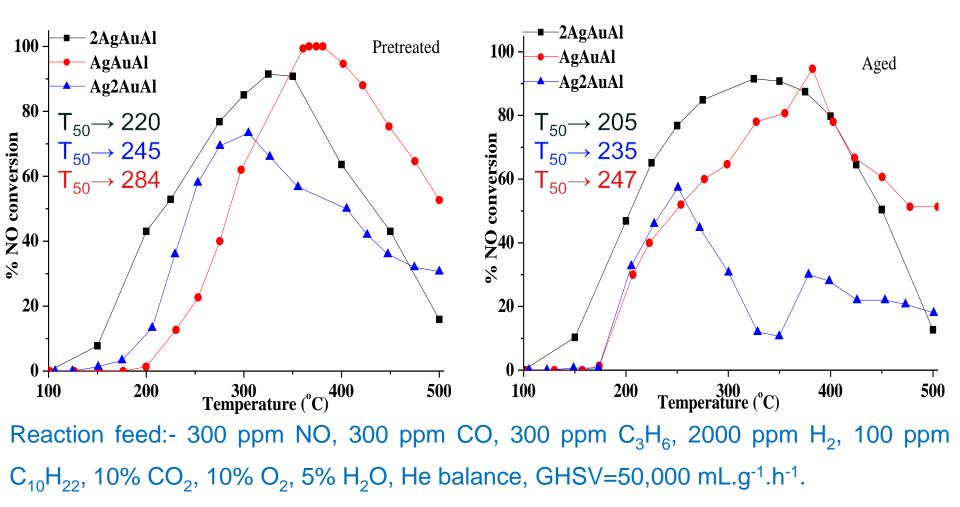
#### **Test Procedure**

- 1. Heat the catalyst powder in hydrogen at 250 °C for 12 h
- 2. Cool to room temperature
- Pass simulated exhaust gas composition on catalyst bed from 80-500 °C @ heating rate of 2 °C/min and monitor the conversion of NO to N<sub>2</sub> (termed as fresh catalyst)
- 4. Pass the exhaust feed on catalyst bed at 800 °C for 8 h
- 5. Cool to 80 °C
- Again pass simulated exhaust gas composition on catalyst bed from 80-500 °C @ heating rate of 2 °C/min and monitor the conversion of NO to N<sub>2</sub> (termed as aged catalyst)

Simulated exhaust gas composition:

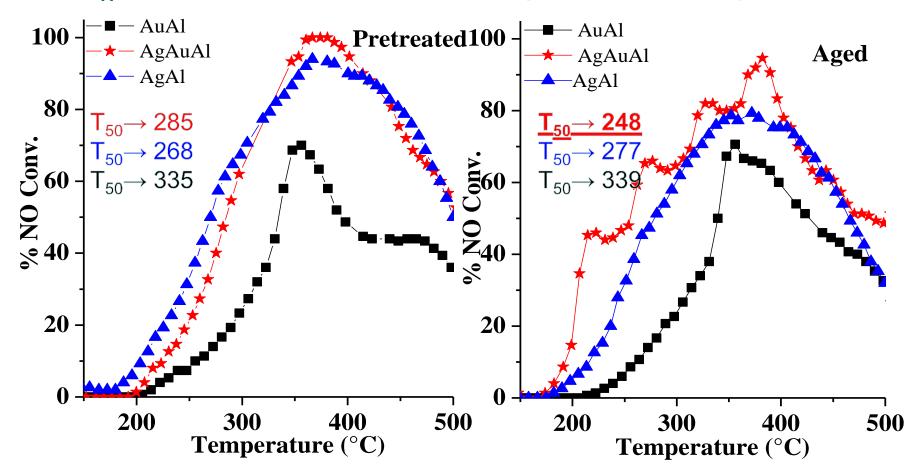
300 ppm NO, 300 ppm CO, 300 ppm  $C_3H_6$ , 2000 ppm  $H_2$ , 100 ppm  $C_{10}H_{22}$ , 10% CO<sub>2</sub>, 10% O<sub>2</sub>, 5% H<sub>2</sub>O, He balance. GHSV = 50,000 mL.g<sup>-1</sup>.h<sup>-1</sup>

### SCR activity of different AgAu catalysts



Promising NO conversion to N<sub>2</sub> in a wide temperature range of 200-500 °C Catalyst activity improved after ageing – treating under simulated exhaust composition at high temperature

#### DeNO<sub>x</sub> activity comparison of AgAuAI with AgAI and AuAI

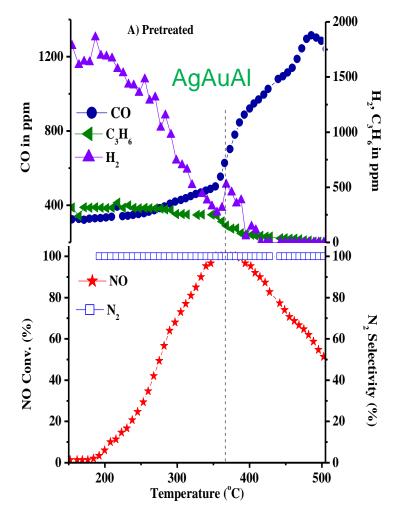


Reaction feed:- 300 ppm NO, 300 ppm CO, 300 ppm  $C_3H_6$ , 2000 ppm  $H_2$ , 100 ppm  $C_{10}H_{22}$ , 10%  $CO_2$ , 10%  $O_2$ , 5%  $H_2O$ , He balance, GHSV=50,000 mL.g<sup>-1</sup>.h<sup>-1</sup>.

After aging activity of AgAuAl increased Almost 50% NO conversion even at 250 °C Activity of bimetallic Ag-Au/Al<sub>2</sub>O<sub>3</sub> is better than only Ag/Al<sub>2</sub>O<sub>3</sub> or Au/Al<sub>2</sub>O<sub>3</sub>

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#### Correlation of deNO<sub>x</sub> activity with measurement of CO, H<sub>2</sub> & C<sub>3</sub>H<sub>6</sub>



- $\circ$  NOx conversion to only N<sub>2</sub>
- No N<sub>2</sub>O formation
- Under reaction condition
  reforming takes place forming
  hydrogen in situ
- In situ formed hydrogen helps for low temperature reduction of NO
- Significant NO reduction even at high temperature (500 °C)

Reaction feed:- 300 ppm NO, 300 ppm CO, 300 ppm  $C_3H_6$ , 2000 ppm  $H_2$ , 100 ppm  $C_{10}H_{22}$ , 10% CO<sub>2</sub>, 10% O<sub>2</sub>, 5% H<sub>2</sub>O, He balance, GHSV=50,000 mL.g<sup>-1</sup>.h<sup>-1</sup>. High DeNOx activity at high temperature <sup>8</sup>

# Summary

- HC-SCR promising for small passenger cars & small tempo type vehicles
- ✓ Unburnt HC or little fuel penalty needed compared to noble metal cost in LNT & catalyst replacement
- Alumina as support hence present alumina wash-coated monoliths can be good starting point
- No ready-made technology available but promising lab scale results available

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Thank You